

2.5.1 Mechanism of internal assessment is transparent and robust in terms of frequency and mode

The following are glimpses from various departments showcasing the transparency of the internal assessment as well as its robustness in terms of frequency and mode:

Date: 03.12.2022

B Sc 5th Semester Internal Examination 2022

Subject: Physics (Honours)

Time : 1 pm – 2 pm


Venue: Seminar Room (Room No. 206)

Sl. No.	Paper	Date	Invigilators
1	Nuclear & Particle Physics (DSE-1)	06.12.2022	Dr. Shilpi S Mandal Dr. J. K. Majhi
2	Atomic & Molecular Spectroscopy (DSE-2)		
3	Quantum Mechanics (CC-11)	07.12.2022	Dr. Shrabani Mondal Dr. Abhik Ghosh
4	Thermal Physics-II (CC-12)		

RKR
HOD

Department of Physics,
B. B. College, Asansol-3

Head
Department of Physics (UG & PG)
B.B. College, Asansol-713303 (W.B)



Date: 17.12.2022

B Sc 3rd Semester Internal Examination 2022

Subject: Physics (Generic)

Time : 12.00 noon – 12.30 pm

Date: 22.12.2022


Venue: Seminar Room (Room No. 206)

Sl. No.	Paper
1	Thermal Physics and Statistical Mechanics (GE-III)

RKR
HOD

Department of Physics,
B. B. College, Asansol-3

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B.B. College, Asansol-713303 (W.B)



Internal exam notice provided by Department of Physics



DEPARTMENT OF PHYSICS (PG & UG)
BANWARILAL BHALOTIA COLLEGE
Asansol — 713 303, West Bengal, INDIA
Government Sponsored with Post-Graduate Faculty

Date: 13.06.2023

NOTICE

The students of B.Sc. Semester IV (Hons) having subject Physics are hereby informed that the University Practical Examination will be held according to the following schedule in the department of Physics.

Date	Roll nos.	Time	Paper	Examiner
21/06/2023	102211220018,102211220024,102211220053,102211220075,102211220092,102211220111	11 am -1pm	IX	KM
	102211220114,102211220117,102211220149,102211220157,102211220169,102211220171, 102211220175	11 am -1pm	X	KM (Jr)
	102211220018,102211220024,102211220053,102211220075,102211220092,102211220111	2.00pm-4.00pm	X	KM (Jr)
22/06/2023	102211220114,102211220117,102211220149,102211220157,102211220169,102211220171, 102211220175	2.00pm-4.00pm	IX	KM
	102211220185,102211220189,102211220236,102211220248,102211220251,102211220307,102211220330	11 am -1pm	IX	KKD
	102211220346,102211220362,102211220366,102211220370,102211220371,102211220387,102211220468	11 am -1pm	X	KM
	102211220185,102211220189,102211220236,102211220248,102211220251,102211220307,102211220330	2.00pm-4.00pm	X	KM
	102211220346,102211220362,102211220366,102211220370,102211220371,102211220387,102211220468	2.00pm-4.00pm	IX	KKD
	102211220346,102211220362,102211220366,102211220370,102211220371,102211220387,102211220468	2.00pm-4.00pm	IX	KKD

Copy to:

1. Dr. Somnath Bhattacharya, External Examiner, Durgapur Govt. College, Durgapur
2. Dr. K. Mukherjee, Internal Examiner, B.B. College, Asansol
3. Dr. K.K. Dey, Internal Examiner, B.B. College, Asansol
4. Dr. J.K. Majhi, Internal Examiner, B.B. College, Asansol
5. Sri. K.Maji, Internal Examiner, B.B. College, Asansol



[Signature]
for Head
Department of Physics
B.B.College Asansol
Head
Department of Physics (UG & PG)
B.B. College, Asansol-713303 (W.B)



DEPARTMENT OF PHYSICS (PG & UG)
BANWARILAL BHALOTIA COLLEGE
Asansol — 713 303, West Bengal, INDIA
Government Sponsored with Post-Graduate Faculty

Date: 13.06.2023

NOTICE

The students of B.Sc. Semester IV (Hons) having subject Physics are hereby informed that the University Practical Examination will be held according to the following schedule in the department of Physics.

Date	Roll nos.	Time	Paper	Examiner
23/06/2023	102211220018,102211220024,102211220053,102211220075,102211220092,102211220111	11 am -1pm	SEC	JKM
	102211220114,102211220117,102211220149,102211220157,102211220169,102211220171, 102211220175	2 am -4pm	SEC	KM
24/06/2023	102211220185,102211220189,102211220236,102211220248,102211220251,102211220307,102211220330	11 am -1pm	SEC	JKM
	102211220346,102211220362,102211220366,102211220370,102211220371,102211220387,102211220468	2pm -4pm	SEC	KKD

Copy to:

1. Dr. Somnath Bhattacharya, External Examiner, Durgapur Govt. College, Durgapur
2. Dr. K. Mukherjee, Internal Examiner, B.B. College, Asansol
3. Dr. K.K. Dey, Internal Examiner, B.B. College, Asansol
4. Dr. J.K.Majhi, Internal Examiner, B.B. College, Asansol



[Signature]
for Head
Department of Physics
B.B.College Asansol
Head
Department of Physics (UG & PG)
B.B. College, Asansol-713303 (W.B)

Internal practical exam notice provided by Department of Physics

Banwarilal Bhalotia College
AFFILIATED TO KAZI NAZUL UNIVERSITY, ASANSOL
(GOVT. SPONSORED UG & PG COLLEGE)
ASANSOL - 713303, WEST BENGAL, INDIA

Ref No. _____ Date: 09/08/2023

DEPARTMENT OF CHEMISTRY

Notice
Subject: Schedule of B. Sc. 2nd Semester GE Internal (Theory) Examination 2023

It is hereby notified that 2nd Semester Chemistry GE Internal (Theory) Examination 2023 will be held following the below schedule.

B. Sc.	Date of Examination	Time	Subject	Full marks
Semester - II G.E.	11.08.2023	11.30 am - 12.15 pm	Physical Chemistry & Organic Chemistry	15

[Signature] 09/08/23
Head of Department,
B. B. College
Department of Chemistry
Banwarilal Bhalotia College, Asansol
Wahagan, Asansol-713303

Phone: 0341-2274842
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BANWARILAL BHALOTIA COLLEGE
DEPARTMENT OF COMMERCE
HINDI SHIFT
(Academic year: 2022-2023)

Date: 24-04-2023

Notice

It is hereby informed that Department of Commerce, Hindi Shift, will conduct Internal Exam for 6th semester, B.com (Hons) and B.com (Program) as per the schedule given below. MCQ types exam will be conducted in College's Computer Lab.

Internal Exam Schedule

Date	Time	Course	Papers	Examinees
06.05.2023 (Saturday)	11.00 am	6 th semester, B.com (Hons.) Hindi Shift	- Auditing and Corporate Governance - Indirect Tax Laws - Advanced Cost Accounting.	Students whose name starts from A to M
06.05.2023 (Saturday)	12.30 pm	6 th semester, B.com (Hons.) Hindi Shift	- Auditing and Corporate Governance - Indirect Tax Laws - Advanced Cost Accounting.	Students whose name starts from N to Z
06.05.2023 (Saturday)	2.00 pm	6 th semester, B.com (Program.) Hindi Shift	- Advanced Cost Accounting - Indian Economy - Personal Selling and Salesmanship	All the Students

Note: For "Computerised Accounting", separate internal Assessment in the form of (Lab Note Book: 05 Marks, Viva :10 Marks and Experiment: 15 Marks) will be conducted by Department. Its date and timing will be notified later.

Md Arif Akhter
Asst. Professor and Head
Department of Commerce
Hindi Shift
B.B. College

Internal exam notices by Department of Chemistry and Commerce

Glimpses of evaluated answer scripts and recorded marks by several departments

Class Test Sem 6H Date: 10-04-2023 Time 1h F.M. 15

Questions:

- List the symmetry elements present in BF_3 molecule and PCl_4^+ ion. (2+2)
- What are the conditions that must be satisfied by the symmetry operations in order to form a group? (4)
- Determine the point groups of eclipsed ethane and benzene. (2+2)
- Prove that $S_{2n} = C_n^2$ when n is odd; the symbols have their usual meaning. (3)

Attendance of students:

Sl. No.	Name in Block Letters	Sign.	Marks obtained	Comment
1	SOMITRA GHOSH	Somitra Ghosh	11	
2	ARKA DAS	Arka Das	11	
3	Ananya Sadhu	Ananya Sadhu	11	
4	Nupur Das	Nupur Das	12½	
5	Shreya Chakraborty	Shreya Chakraborty	12	
6	Priya Pal	Priya Pal	12	
7	MERAJUL MAJI	Merajul Maji	4½	
8	Priyam Mishra	Priyam Mishra	11	
9	Subham Paul	Subham Paul	10½	
10	AYAN KONGER	Ayan Konger	10½	
11	MD SOHAIL AKHTAR	Md. Sohail Akhtar	08	
12	CHANDAN MAJI	Chandan Maji	11	
13	BIKRAM KUMAR YADAV	Bikram Kumar Yadav	04	

Signature of the teacher with date of publication of the result of the class test
24-04-2023

Class Test Sem 3H F.M. 10 Time 1h Date: 28.11.2022

1. The dipole moment of o-xylene is 0.693D. Find the dipole moment of toluene. What will be the value in SI unit? (3)

2. Explain the terms: distortion polarization, orientation polarization. (1.5x2 = 3)

3. At 20°C, the refractive index and density of water are 1.333, 0.998 g/cc. Calculate the molar refraction of water. (2)

4. Draw a neat diagram showing the plots of vdw potential $U_v = -A/r^n$ and the Lennard Jones potential $U_L = -A/r^{12} + B/r^6$ for molecular interaction. (2)

Attendance of the students:

Sl. No.	Name	Signature	Marks obtained	Remarks
1	SABYASACHI SINGHA	Sabyasachi Singha	09	
2	ANIRBAN SADHU	Anirban Sadhu	08	
3	TWINKLE MUKHERJEE	Twinkle Mukherjee	7½	
4	SENERITA PADHA	Senenita Padha	9½	
5	TRIPARNA DATTA	Triparna Datta	09	
6	SHREYA GHANTJ	Shreya Ghantj	07	
7	KRISHNA PAL	Krishna Pal	05	
8	ANKITA DAS	Ankita Das	3½	
9	MOUSUMI MISRA	Mousumi Misra	04	
10	SHRUTI ROY	Shruti Roy	4½	
11	ABASI GOSWAMI	Abasi Goswami	05	
12	ANINDITA ROY	Anindita Roy	05	
13	SUBHADIP MAJI	Subhadip Maji	6½	
14	SUDIPTA MUKHERJEE	Sudipta Mukherjee	7½	
15	MAYANK SINHA	Mayank Sinha	1½	
16	RACHIT SHARMA	Rachit Sharma	02	

Signature of the Teacher: 02.12.2022

Class Test Semester 1 Time 1h F.M. 10 Date: 15.12.2022

Questions:

- Draw the molecular orbital diagram of O_2 and hence find out the bond order of O_2 , O_2^+ , O_2^- , O_2^{2+} . [2+2]
- Write down the three conditions of LCAO. [3]
- Draw the pictures of the molecular orbitals resulting in the head on interaction of $2p_x$ orbital of Be and $2p_x$ orbitals of two F atoms in BeF_2 molecule. Consider x-axis to be the bond axis. Comment on their gerade and ungerade character. [2+1]

Student's attendance:

Sl. No.	Name	Signature	Marks obtained	Remarks, if any
1	Bibek Chauhan	Bibek Chauhan	06	
2	Debjit Roy	Debjit Roy	3½	
3	Indrani Pattanayak	Indrani Pattanayak	05	
4	Koyel Mukherjee	Koyel Mukherjee	05	
5	Muskan Kumari	Muskan Kumari	6½	
6	Pushon Biswas	Pushon Biswas	8½	
7	Rahul Maji	Rahul Maji	5½	
8	Rahul Mondal	Rahul Mondal	6½	
9	Rima Pal	Rima Pal	04	
10	Rima Panda	Rima Panda	05	

Teacher's Signature with date:

11. Sayandeep Das ✓ Sayandeep Das 05

12. Sujan Patra ✓ Sujan Patra 4½

13. Akshat Thakur ✓ Akshat Thakur 04

14. Soumya Ghosh ✓ Soumya Ghosh 4½

Priyam Mishra

Photochemistry

26/04/23 (6/10)

1) Principle laws of photochemistry are:

Grotthuss-Draper law:

It is also known as first law of photochemistry. It states that, only the radiations which are absorbed by the reactant or molecules of the system will be efficient to produce any chemical changes. The radiations which are transmitted or reflected are unable to produce chemical change.

$$I_0 \xrightarrow{\text{absorption}} I \quad (I_{\text{abs}} = I_0 - I)$$

ark Einstein law:

This law states that in a photochemical reaction each molecule of reacting substance absorbs a single photon of radiation, causing the reaction and activated to form product.

$$\text{Reactant} \xrightarrow{h\nu} \text{A}^* \rightarrow \text{B (product)}$$

2) Quantum yield (ϕ) = $\frac{\text{no of molecules reacts per sec}}{\text{no of quanta absorbed per sec}}$

$$= \frac{0.002 \times 6.02 \times 10^{23}}{1200 \times 2 \times 10^{16}}$$

$$= \frac{2 \times 6.02 \times 10^{20}}{12 \times 2 \times 10^{18}}$$

$$= 0.5017 \times 10^2$$

Name: Sneha Dutta
 Sem: 6th
 Subject: Photochemistry

9/10 Roy 26/04/23

1) principal laws of photochemistry are as follows:

(i) Grothius-Draper law: This law states that only absorbed light by a molecule can cause any physical or chemical change.

(ii) Stark-Einstein law: This law states that one quantum of light is absorbed by a molecule of absorbing or reacting substance that disappears.

2) No. of molecules reacting in 1200 seconds
 $= 0.002 \times 6.02 \times 10^{23}$

\therefore No. of molecules reacting per second $= \frac{0.002 \times 6.02 \times 10^{23}}{1200}$

$\therefore \phi$ (Quantum yield) $= \frac{\text{No. of molecules reacting}}{\text{No. of photons absorbed}}$

$$= \frac{0.002 \times 6.02 \times 10^{23}}{1200} \times \frac{1}{2 \times 10^{16}}$$

$$= 50.16$$

3) (b) A photoluminescence where the light lasts a long time.

4) The energy difference between the vibrational bands are supposed to be ΔE_1 and in the absorption spectra and ΔE_2 in the emission.

Topic: Polymer Chemistry
Inorganic Chemistry-V
BSCHEM C601 (Unit-IV)

F.M.-10 Time-1hr.
 Mem Sr.
 Date: 26/04/23
 Ananya Sadhu.

1) $F_2C=CF_2$ is monomer of 9/10
 (a) teflon (b) orlon (c) polythene (d) nylon-6

2) Natural rubber is a polymer of 1
 (a) Styrene (b) ethene (c) butadiene (d) isoprene

3) The bakelite is prepared by the reaction between
 (a) Phenol and formaldehyde (b) tetramethylene glycol
 (c) urea and formaldehyde (d) ethylene glycol. 1

4) Among the following substituted silanes the one which will give rise to cross linked silicone polymer on hydrolysis is -
 (a) R_4Si (b) $RSiCl_3$ (c) R_2SiCl_2 (d) R_3SiCl 1

5) The two monomers for the synthesis of Nylon 6,6 are 1
 (a) $HOOC(CH_2)_6COOH$, $H_2N(CH_2)_6NH_2$ (b) $HOOC(CH_2)_4COOH$, $H_2N(CH_2)_6NH_2$
 (c) $HOOC(CH_2)_4COOH$, $H_2N(CH_2)_6NH_2$ (d) $HOOC(CH_2)_6COOH$, $H_2N(CH_2)_4NH_2$

6) Caprolactam is used for the manufacture of 1
 (a) teflon (b) terylene (c) nylon 6,6 (d) nylon 6

7) Let us suppose that there are 100 polymer molecules with molecular weight 2000, 200 polymer molecules with molecular weight 5000 and 300 molecules with molecular weight 10000. Then calculate the Numer average M.W (M_n) and weight average M.W (M_w)? 1

Chandan Nayak
 6th sem (B.Sc III - Chemistry)
 KNU20102004686

7/2
 10
 Roy 26/04/23

Photochemistry / Polymer chemistry - class test

1) The principal laws of photochemistry

(i) Grothius-Draper law -
 It was realized that all incident light not effective in bringing that change in system. The first law of photochemistry states that only light which is absorbed by system can cause a chemical change.

(ii) Einstein law -
 One quantum of light is absorbed per molecule of any absorbing and reacting substances that disappears.

2) 2.0×10^6 quanta

Total no. of moles react 0.002 moles in 1200s

\therefore Total no. of molecules react per second $= \frac{0.002 \times 6.02 \times 10^{23}}{1200}$

$$= 1.003 \times 10^{18}$$

\therefore Quantum yield $= \frac{1.003 \times 10^{18}}{2.0 \times 10^6}$

Md. Saikat Akhter
 KNU20102004584
 26/04/23

10/10

7) Given: $n_1 = 100$, $n_2 = 200$, $n_3 = 300$
 $M_1 = 2000$, $M_2 = 5000$, $M_3 = 10000$

\therefore We know, Number average M.W (M_n) $= \frac{n_1M_1 + n_2M_2 + n_3M_3}{n_1 + n_2 + n_3}$

$$= \frac{100 \times 2000 + 200 \times 5000 + 300 \times 10000}{100 + 200 + 300}$$

$$= \frac{200000 + 1000000 + 3000000}{600}$$

$$= \frac{4200000}{600} = 7000$$

\therefore Weight average Molecular weight (M_w) $= \frac{n_1M_1^2 + n_2M_2^2 + n_3M_3^2}{n_1M_1 + n_2M_2 + n_3M_3}$

$$= \frac{100 \times (2000)^2 + 200 \times (5000)^2 + 300 \times (10000)^2}{100 \times 2000 + 200 \times 5000 + 300 \times 10000}$$

$$= \frac{4 \times 10^8 + 5 \times 10^9 + 3 \times 10^{10}}{42 \times 10^5}$$

$$= \frac{3.54 \times 10^{10}}{42 \times 10^5} = \frac{354000000}{420000}$$

Notices of project reports and seminar presentations by several departments

SEC-II (BSc Physics Honours 4th Semester Project Distributions)

Basic Instrumentation Skills (BSCHPHSSEC401)

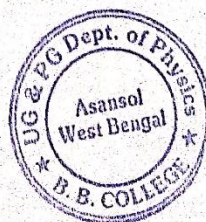
Experiment/Project	Teacher	Students
Measurement of Voltage, Frequency, Time Period and Phase angle.	Dr. P. Ghosh	1. Ankita Banerjee 2. Debajyoti Mondal 3. Debangshu Chakraborty
Measurement of rise, fall and delay time using CRO	Dr. K. Mukherjee	4. Debashish Gope 5. Debdeep Das 6. Debsmita Chatterjee
Measurement of R, L and C using LCR bridge or universal bridge	Dr. J. K. Majhi	7. Deep Das 8. Jagannath Ray 9. Jit Dey
Working principle of function generator	Dr. R. K. Roy	10. Kirti Kumari Yadav 11. Neha Sah 12. Niraj Kumar Paswan
Working Principle of Q meter with Block diagram	Dr. K. K. Dey	13. Priyangshuk Chakraborty 14. Priyanshu Mandal 15. Puja Harijan
To draw the Lissajous figure using CRO	Sri K. Maji	16. Rudra Sinha 17. Rupa Sadhu 18. Sagar Mondal
Working principle of Digital meters (Volt meter, ammeter etc.)	Dr. S. Mondal	19. Sahil Sarfaraz Khan 20. Sangita Maddi 21. Santanu Chakraborty
Working principle of AC milli voltmeter	Dr. S. S. Mandal	22. Satyadeep Modak 23. Sayan Shib Das 24. Seema Dey
Working principle and application of transformer	Dr. A. Ghosh	25. Somnath Chattaraj 26. Sourik Dey 27. Subhodeep Mukherjee 28. Konika Roy
To observe the loading effect of a multimeter while measuring voltage across a low resistance and high resistance	Dr. A. Biswas	29. Sujoy Banerjee 30. Tanima Bandyopadhyay 31. Wilson Raj

RKR 02.05.2023

HOD

Department of Physics, B. B. College, Asansol.

Head
Department of Physics (UG & PG)
B.B. College, Asansol-713303 (W.B)





DEPARTMENT OF PHYSICS (PG & UG)
BANWARILAL BHALOTIA COLLEGE
Asansol — 713 303, West Bengal, INDIA
Government Sponsored with Post-Graduate Faculty

Date: 28/08/2023

NOTICE

The students of M.Sc. Semester II are hereby informed that seminar presentation will be held according to the following schedule in the Department of Physics.

Date	Registration nos.	Time
	102223320	
01/09/2023	001, 002, 004, 005, 006, 007 008, 009, 010, 011, 012, 013	11 am -1 pm

B.B. College, Asansol (Department of Physics)

B.Sc Semester-III SEC project Distribution 2022

Electrical Circuit Network Skills

Name of the supervisors	Project Title	Name of Students
Dr. P. Ghosh	Design and study of IC regulated 5V D.C power supply	Rupa Sadhu Seema Dey, <i>Sayanshit Das</i>
Dr. K. Mukherjee	To study the resistors, capacitors and inductors to different types of signals	Ankita Banerjee Debsmita Chatterjee <i>Wilson</i>
Sri. K. Maji	Digital multi-meter and its working principle	Pratichhe Mondal Priyanshu Mondal Satyadeep Modak
Dr. S.Mandal	To study the basic electricity principles	Deep Das Sagar Mondal Debojyoti Mondal
Dr. R.K. Roy	Conversion of A.C. to D.C by bridge rectifier with capacitor filter	Jagannath Ray Jit Dey Rudra Sinha
Dr. J. K. Majhi	Conversion of A.C to D.C by full wave rectifier	Priyanshuk Chakraborty Tanima <i>Sujoy</i> Bandopadhyay, <i>Banijet</i>
Dr. A. Ghosh	Conversion of Ammeter to Voltmeter	Sourik Dey Somnath Chattaraj
Dr. S.S. Mandal	Conversion of Voltmeter to Ammeter	Debashis Gope Subhodeep Mukherjee Neha Sah
Dr. K.K. Dey	Construction of house hold electrical wiring (single-phase)	Sangita Maddi Kirti Kumari Yadav Puja Harijan

RKR 08.09.2022


H.O.D

Department Of Physics

Department of Physics (UG & PG)
B.B. College, Asansol-713303 (W.B)

Project reports submitted by students

KAZI NAZRUL UNIVERSITY




PROJECT REPORT ON WOMEN EMPOWERMENT
B.A.HONS 6TH SEMESTER EXAMINATION

SUBMITTED BY
NAME:- SUSMITA KUMARI KOIRI
REGISTRATION NO : KNU20116002112
ROLL NO.:- 1162006121034102
SUBJECT TYPE :- CC-14
SUBJECT CODE:- BAHPLSC602
SUBMITTED TO - PAWAN GURUNG & SANTOSH BHAGAT
DEPARTMENT OF POLITICAL SCIENCE
(HINDI SHIFT)
BANWARILAL BHALOTIA COLLEGE
USHAGRAM, ASANSOL-713303
2020-2023

BANWARILAL BHALOTIA COLLEGE
(Kazi Nazrul University)

Name : Saraswati Kumari
Course : B.A (Hons) Political Science
Roll No : HAH-547/20
Session : 2020-2023
Mobile : 9907345984
Address : Dhamjour Rangpasta A/BType 37/6
E-mail : SaraswatiKumari2650@gmail.com
Application NO : BBCOLL-1580125/20
TOPIC : Problems, challenges and Security concerns faced by transgender People.
Submitted to:- Pawan Gurung

Kazi Nazrul University
Banwarilal Bhalotia College




Quadrilateral Security Dialogue

Name : Puja Kumari
Reg. no: KNU20116001728
Course name: Project
Course code: BAHPLSC602
Department: Political science

Submitted to
Pawan Gurung
&
Santosh Bhagat

Submitted by
Puja kumari

काजी नजरूल विश्वविद्यालय
बनवारीलाल भलोदिया महाविद्यालय, आसनसोल



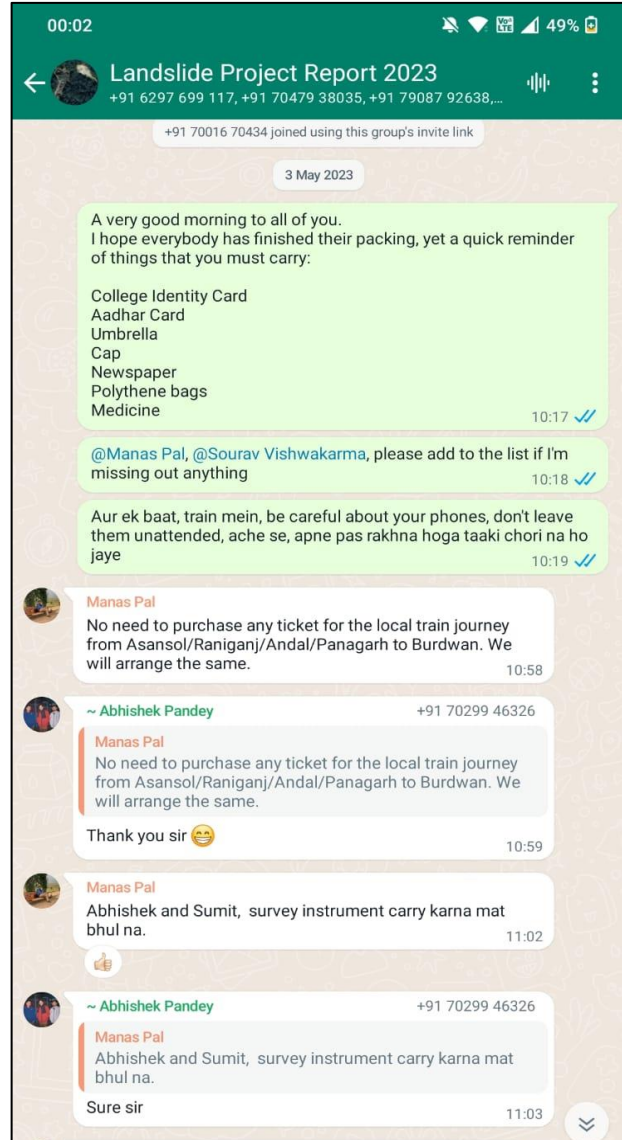
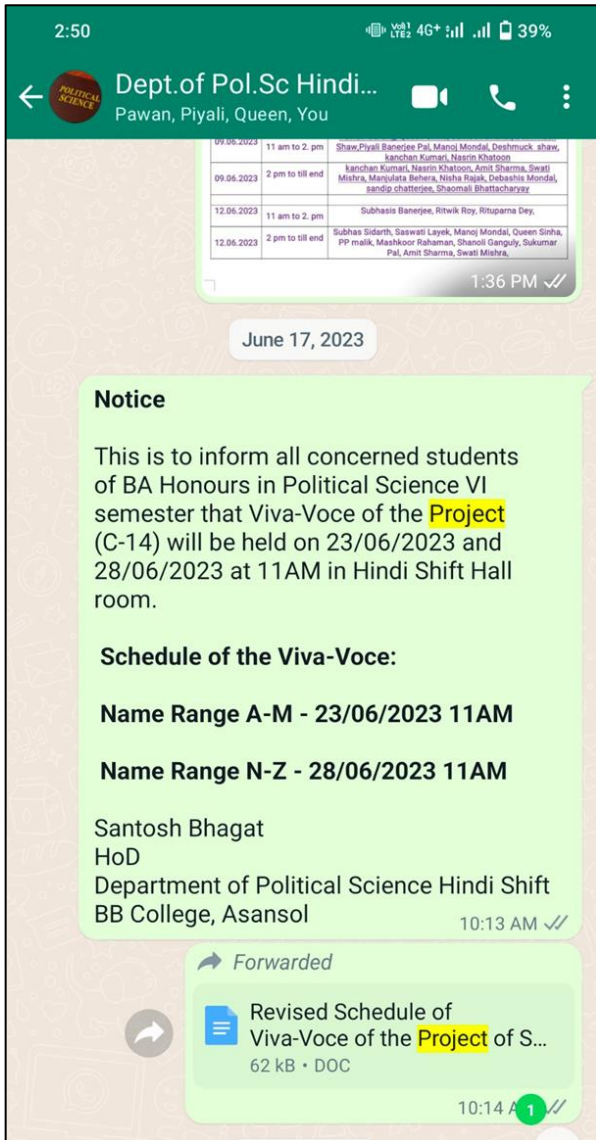
संयुक्त राष्ट्र विकास कार्यक्रम परियोजना

नाम :- आर्यन शर्मा
अनुक्रमांक :- 1162005121034017
पंजीकरण संख्या:- KNU20116002169
पाठ्यक्रम प्रकार:- C
पाठ्यक्रम संहिता :- BAHPLSC602
विभाग:- राजनीति विज्ञान (हिंदी पारी)
सत्र :- २०२३-२०२३

आर्यन शर्मा
द्वारा प्रस्तुत किया गया।

सहायक प्राध्यापक पवन गुरुंग
को प्रस्तुत किया गया।

WhatsApp group created by respective departments for ease of communication



Glimpses of assignments

